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$$p = f(t) t + O(t)^2: (1.2)$$

Therefore, the contribution to the average electron momentum from the electrons that do not collide is  $p(t+t) = (1-t)(p(t) + f) t + O(t)^2: (1.3)$  Note that the contribution to the average momentum from electrons which have collided will be of order  $(t)^2$  and therefore negligible.

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 $\text{Fe (s)} + \text{FeSO}_4 \text{ (aq)} \rightarrow \text{no reaction.}$   
 $\text{Fe (s)} + \text{Al}_2(\text{SO}_4)_3 \text{ (aq)} \rightarrow \text{no reaction.}$   
 $\text{Fe (s)} + \text{ZnSO}_4 \text{ (aq)} \rightarrow \text{no reaction.}$   
No reaction occurs when any  
of the metals from zinc,  
iron, aluminium, and copper  
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